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(54) Title: AQUEOUS BRANCHED POLYMER DISPERSION			
(57) Abstract An aqueous coating composition comprising a dispersed graft copolymer characterized by macromonomers attached at a terminal end thereof to a polymeric backbone which is acid functional, wherein the functional groups have been neutralized. Such compositions provide improved properties for an automotive finish.			

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TITLE
AQUEOUS BRANCHED POLYMER DISPERSION

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Technical Field

This invention relates to an improved aqueous composition for coating a variety of substrates. In particular, this invention is directed to a waterborne coating composition comprising a graft copolymer, which is a branched copolymer having neutralized carboxylic functionality.

10

BACKGROUND OF THE INVENTION

Automobiles and trucks receive exterior finishes for several well known reasons. First, such finishes provide barrier protection against corrosion. Second, consumers prefer an exterior finish having an attractive aesthetic finish, including high gloss and excellent DOI (distinctness of image).

A typical automobile steel panel or substrate has several layers of finishes or coatings. The substrate is typically first coated with an inorganic rust-proofing zinc or iron phosphate layer over which is provided a primer which can be an electrocoated primer or a repair primer. Optionally, a primer surfacer can be applied to provide for better appearance and/or improved adhesion. A pigmented basecoat or colorcoat is next applied over the primer. A typical basecoat or colorcoat comprises a pigment, which may include metallic flakes in the case of a metallic finish. In order to protect and preserve the aesthetic qualities of the finish on the vehicle, it is well known to provide a clear (unpigmented) topcoat over the colored (pigmented) basecoat, so that the basecoat remains unaffected even on prolonged exposure to the environment or weathering.

Coating compositions comprise one or more film-forming polymers. Most commonly, linear polymers are employed that cure, upon application, by reaction with crosslinking agents. However, the use of non-linear graft copolymers has been disclosed. For example, U.S. Patent No. 4,801,653 to Das et al. describes the use of hydroxy functional graft copolymers. Das et al. disclose grafting by a condensation reaction between epoxy groups of a glycidyl ester, contained in an acrylic polymer, and carboxy

groups on at least a portion of vinyl monomers which are polymerized in the presence of the acrylic polymer.

In preparing graft polymers in general, various living polymerization methods have been disclosed for obtaining functional ended

- 5 polymers by selective termination of living ends. Such functionally ended polymers may subsequently be attached to another polymer, that is, as so-called macromonomer "arms" on a polymeric backbone to form a comb graft copolymer. Webster, in "Living Polymerization Methods," 251
10 SCIENCE 887 (22 February 1991) generally discloses living polymerization methods for preparing architectural forms of polymers, including graft and comb copolymers.

Graft copolymers containing carboxyl groups and the preparation of these polymers is shown in Japanese Laid Open Patent Application (Kokai) No. 1-182304 dated July 20, 1989. This reference shows 15 graft copolymers that have carboxyl groups based on acrylic and methacrylic acid in their side chains that have hydrophilic properties and teaches the use tertiary alcohol-based ester units of acrylic or methacrylic acid to form a macromonomer which is used to form a graft copolymer and then is hydrolyzed to form carboxylic acid groups on the polymer. The process 20 taught by the reference is an inefficient process which does not form pure graft copolymer but results in a mixture of graft copolymer and low molecular weight components that are detrimental to pigment dispersions formed from the graft copolymer and finishes formed from such a composition.

- 25 U.S. Patent No. 4,680,352 to Janowicz et al. and U.S. Patent No. 4,722,984 to Janowicz disclose the use of cobalt (Co) chelates as chain transfer agents in free radical polymerization. The latter patent discloses that macromonomers prepared by cobalt chain transfer can be polymerized to produce graft copolymers which are useful in coating and molding resins, 30 including high solid finishes and aqueous or solvent based finishes, although the use of such polymers have so far found only limited use in the finishes area. The use of chain transfer agents to prepare dispersed polymers used in automotive coatings is disclosed in U.S. Patent No. 5,010,140.

The present invention relates to aqueous coating compositions.

- 35 The evolution of environmental regulations has led to the need for products with lower volatile organic content (VOC). Water dispersible polymers are

well known in the art and have been used to form water-based coating compositions, pigment dispersions, adhesives and the like. Much of the prior art, however is directed to either latex or water reducible linear polymers. Latex-based systems tend to have problems with coalescence. Linear water
5 reducible polymers are difficult to process because of high molecular weight.

BASF EP 0363723 describes an acrylic copolymer dispersion for use in an OEM clear coat to be crosslinked with a melamine formaldehyde resin. The linear acrylic copolymer is prepared in a solvent in a two-stage process where the hydrophilic part (acid-functional monomer) is
10 concentrated in one of the two stages. The overall copolymer is afterwards neutralized with an amine and dispersed in water. The difference from a one-stage product is that the solids/viscosity relation is more favorable. A disadvantage of this technology is the fact that the hydrophilic part needs to be over 60% of acid-functional monomer which could give problems in
15 humidity resistance.

Bayer patents EP 0218906 and EP 0324334 describe the synthesis of hydroxy-acid functional acrylic copolymers prepared in solution before neutralizing with an amine and dispersing in water. Bayer EP 0334032 describes the synthesis of an acid functional urethane oligomer which is used
20 to stabilize a WB acrylic copolymer dispersion. AKZO US 5,098,947 describes urethane modified acrylic copolymer dispersions for waterborne coatings.

Regarding the presently claimed waterborne composition, branched copolymers allow for higher molecular weight polymers at lower
25 viscosity compared to linear polymers of the same molecular weight. Such branched copolymers also allow for segmentation of the copolymer into hydrophobic and hydrophilic portions for improved coating properties. Accordingly, the branched copolymers of the present invention are hydrosols that form a dispersed phase at high pH and high water content while
30 forming a solution polymer at low pH and/or high organic solvent/low water content. This ease of inversion can provide coatings with better pin-hole resistance among other desirable properties. Applicants have found that coatings made from such aqueous branched copolymers are hard, water and humidity resistant, and show excellent humidity characteristics.

SUMMARY OF THE INVENTION

The present invention is directed to a coating composition comprising:

- 5 (a) from about 10 to 90 percent, based on the weight of the binder, of a graft copolymer having a weight average molecular weight of 2000-100,000 comprising:
- 10 (i) 5 to 95 percent by weight of the graft polymer of a polymeric backbone comprising polymerized ethylenically unsaturated monomers of which 2 to 35% by weight have carboxylic functionality; and
- 15 (ii) 95 to 5 percent, by weight of the graft polymer, of macromonomers attached to said polymeric backbone at a single terminal point of each macromonomer, said macromonomers comprising polymerized ethylenically unsaturated monomers containing less than 1% carboxylic functionality and having a weight average molecular weight of about 1,000-30,000;
- 20 (b) 0 to 40 percent of a crosslinking agent which, when the composition is cured, can react and crosslink with said crosslinking functionality in said graft polymer; and
- 25 (c) from 10% to 90% by weight, based on the weight of the composition, of an aqueous carrier comprising at least 50% water; wherein said carboxylic functionality has been at least partially neutralized with an amine or other neutralizing agent to form a stable dispersion or solution in water.
- 30 The above-defined graft copolymer may be present in the binder in a substantial amount or as the primary (majority) component or even as the only film-forming polymer with the exception of crosslinking agents and rheology controllers. This graft copolymer may also be employed together with other curable components, including a latex or a curable linear film-forming acrylic, polyester, or polyester urethane polymer, in various proportions.

35 The present composition is especially useful for finishing the exterior of automobiles and trucks and parts thereof. The present composition, depending on the presence of pigments and other conventional components, may be used as a primer, primer surfacer, basecoat, and/or clearcoat. The invention also includes a process for coating a substrate with

the above coating composition. The claimed composition further includes a substrate having adhered thereto a coating according to the above composition. The graft copolymer and the process for making the graft copolymer are also part of this invention.

5

DETAILED DESCRIPTION OF THE INVENTION

Waterborne curable compositions comprising a blend of a dispersed graft copolymer, which contains active hydrogens, and a curing agent are disclosed. The graft copolymer is prepared from an acrylic copolymer macromonomer comprising polymerizable alpha-beta ethylenically unsaturated monomers and a weight average molecular weight (MW) of 1000-30,000, preferably 6000 to 15,000. About 5-95% (by weight), preferably 30-70%, of the macromonomer is copolymerized with 95-5%, preferably 70-30%, of a blend of other alpha, beta-ethylenically unsaturated monomers, at least 2%, preferably 2-30% by weight, most preferably 3-15%, of which have a carboxylic acid functionality, to form a branched or graft copolymer with a MW of 2000-100,000, preferably 5000-40,000, which after neutralizing with an amine or other neutralizing agent can be dispersed in water. (All molecular weights herein are based on GPC using a polystyrene standard.)

It has been found that improved aqueous or waterborne coating systems are obtained by using these graft copolymers in combination with a curing agent, suitably an amine compound as a crosslinker. These polymers can be utilized as film-formers either alone or in combination with a latex or other polymers. Such compositions have the advantage of providing excellent coating properties desirable for automotive finishes.

The waterborne coatings of the present invention comprise an acrylic-based binder system in an aqueous base. This binder system comprises, in its overall concept, a water soluble or dispersible graft acrylic copolymer which is formed by free-radical initiated copolymerization of 5-95% (weight) alpha-beta unsaturated monomers in the presence of 95-5% of an acrylic macromonomer. The acrylic macromonomer is anionic in character with a preferred average number molecular weight (MN) of between 500 to 20,000 and containing less than about 1% of an acid functional alpha-beta unsaturated monomer, so that the backbone remains relatively hydrophilic and the macromonomer side chains remain relatively

hydrophobic, following at least partial neutralization of the carboxyl groups with, for example, an amine. These acrylic resins form stable solutions or dispersions in water, typically as a dispersed polymer having an average particle size diameter of 10 nm to 1 micron, preferably 20 to 400 nm.

5 In general, the total polymeric and oligomeric components of a coating composition are conventionally referred to as the "binder" or "binder solids" and are dissolved, emulsified or otherwise dispersed in an aqueous liquid carrier. The binder solids generally include all the normally solid polymeric components of the composition. Generally, catalysts, pigments, or
10 chemical additives such as stabilizers are not considered part of the binder solids. Non-binder solids other than pigments usually do not amount for more than about 10% by weight of the composition. The total coating composition of the present invention suitably contains about 10-70%, more typically 15-50% by weight of binder, and about 30-90%, more typically
15 50-85% by weight, of an aqueous carrier. The carrier is at least 50% water, preferably 75 to 95% water.

The present composition suitably comprises about 10 to 90 percent, preferably 20 to 60%, based on the weight of the binder, of the specified graft polymer.

20 The side chains of the graft copolymer are preferably hydrophobic relative to the backbone and therefore contain less than 1% by weight, preferably essentially zero percent by weight, based on the weight of the graft copolymer, of polymerized ethylenically unsaturated acid-functional monomers which are listed hereinafter. The side chains contain polymerized hydrophobic monomers such as alkyl methacrylates and acrylates, cycloaliphatic methacrylates and acrylates and aryl methacrylates and acrylates as are listed hereinafter and also may contain up to 30% by weight, based on the weight of the graft copolymer, of polymerized ethylenically unsaturated non-hydrophobic monomers which may contain functional
25 groups. Examples of such monomers are hydroxy ethyl acrylate, hydroxy ethyl methacrylate, acrylamide, nitro phenol acrylate, nitro phenol methacrylate, phthalimido methyl acrylate, phthalimido methacrylate, acrylic acid, acryloamido propane sulfonic acid, and mixtures thereof.

30 The acrylic macromonomer may be prepared using a free radical initiator in a solvent with a Co (II) or Co (III) chelate chain transfer agent.

The backbone of the graft polymer contains preferable at least 2 percent by weight of an acid functional (neutralized) monomer as, e.g., acrylic acid, methacrylic acid, maleic acid, itaconic acid and the like.

- Methacrylic and acrylic acid are preferred. Other acids that can be used are
5 ethylenically unsaturated sulfonic, sulfinic, phosphoric or phosphonic acid and esters thereof also can be used such as styrene sulfonic acid, acrylamido methyl propane sulfonic acid, vinyl phosphonic or phosphoric acid and its esters and the like.

- The backbone is preferably based on 2-30% methacrylic acid,
10 preferably 3 to 15% and has an MN = 500-30,000. The acid functional groups on the graft copolymer are neutralized with an inorganic base or an amine. The backbone is thus relatively hydrophilic and keeps the graft polymer well dispersed in the resulting coating composition. Of course, relative hydrophobicity or hydrophilicity could be further adjusted by varying
15 the percent of acid and/or hydroxy functional monomers versus more hydrophobic monomers such as 2-ethyl hexyl methacrylate.

- In the preferred embodiments, the waterborne acrylic graft copolymers contain overall (including both backbone and macromonomer arms) about 0 to 40, preferably 5 to 40, and more preferably 10 to 30, parts
20 by weight of hydroxy functional acrylic monomers as, e.g., 2-hydroxyethyl acrylate, 2-hydroxyethyl methacrylate, 2-hydroxypropyl acrylate, 2-hydroxypropyl methacrylate, 2-hydroxybutyl acrylate, and 4-hydroxybutyl acrylate. These hydroxy groups can be used for crosslinking in addition to the acid groups. Hydroxy groups are not necessary when acid groups are the
25 only crosslinking functionality on the copolymer. Hydroxy groups are necessary when the cross-linking agent is melamine or blocked isocyanate.

- As indicated earlier, the graft polymer comprises
macromonomeric side chains attached to a polymeric backbone. Each
30 macromonomer ideally contains a single terminal ethylenically unsaturated group which is polymerized into the backbone of the graft copolymer and typically contains polymerized monomers of esters and/or nitriles and/or amides of methacrylic or acrylic acid or mixtures of these monomers.

- Other polymerized ethylenically unsaturated monomers can be present in the macromonomer and backbone, for example (but not limited
35 to), acrylic and methacrylic acid esters of straight-chain or branched monoalcohols of 1 to 20 carbon atoms. Alkyl acrylates having 1-12 carbons

in the alkyl group can be used such as methyl acrylate, ethyl acrylate, propyl acrylate, isopropyl acrylate, butyl acrylate, pentyl acrylate, hexyl acrylate, 2-ethyl acrylate, nonyl acrylate, lauryl acrylate and the like can be used.

5 Cycloaliphatic acrylates can be used such as trimethylcyclohexyl acrylate, t-butyl cyclohexyl acrylate, cyclohexyl methacrylate, isobornyl methacrylate, 2-ethylhexyl methacrylate, and the like. Aryl acrylates such as benzyl acrylate also can be used.

Ethylenically unsaturated monomers containing hydroxy functionality include hydroxy alkyl acrylates and hydroxy alkyl methacrylates, 10 wherein the alkyl has 1 to 12 carbon atoms. Suitable monomers include hydroxy ethyl acrylate, hydroxy propyl acrylate, hydroxy isopropyl acrylate, hydroxy butyl acrylate, hydroxy ethyl methacrylate, hydroxy propyl methacrylate, hydroxy isopropyl methacrylate, hydroxy butyl methacrylate, and the like, and mixtures thereof. Hydroxy functionality may also be 15 obtained from monomer precursors, for example, the epoxy group of a glycidyl methacrylate unit in a polymer. Such an epoxy group may be converted, in a post polymerization reaction with water or a small amount of acid, to a hydroxy group.

Suitable other olefinically unsaturated comonomers include: 20 acrylamide and methacrylamide and derivatives as alkoxy methyl (meth) acrylamide monomers, such as methacrylamide, N-isobutoxymethyl methacrylamide, and N-methylol methacrylamide; maleic, itaconic and fumaric anhydride and its half and diesters; vinyl aromatics such as styrene and vinyltoluene; and polyethylene glycol monoacrylates and 25 monomethacrylates.

Other functional monomers as itaconic or maleic anhydride, the half ester thereof, acrylonitrile, allylmethacrylate, aceto acetoxyethyl methacrylate, methylacryl amidoglycolate methylether, ethylene urea ethyl methacrylate, 2-acrylamide-2 methyl propanesulfonic acid, trialkoxy silyl 30 ethyl methacrylate, reaction products of mono epoxyesters or monoepoxy ethers with alpha-beta unsaturated acids and reaction products of glycidyl (meth) acrylate with mono functional acids up to 22 carbon atoms.

The above monomers also can be used in the backbone of the graft copolymer.

35 The graft polymer may be prepared by polymerizing ethylenically unsaturated monomers in the presence of macromonomers each

- having a terminal ethylene unsaturation for grafting. The resulting graft polymer can be envisioned as being composed of a backbone having a plurality of macromonomer "arms" attached thereto. In the present composition, both the macromonomer arms and the backbone may have
- 5 reactive functionalities capable of reacting with a crosslinking compound or polymer, although it is optional to have such reactive functionalities only or essentially only or substantially only on the backbone.

It is to be understood that the backbone or macromonomers referred to as having functionality may be part of a mixture of

10 macromonomers of which a portion do not have any functionality or variable amounts of functionality. It is also understood that, in preparing any backbone or macromonomers, there is a normal distribution of functionality.

To ensure that the resulting macromonomer only has one terminal ethylenically unsaturated group which will polymerize with the

15 backbone monomers to form the graft copolymer, the macromonomer is polymerized by using a catalytic chain transfer agent. Typically, in the first step of the process for preparing the macromonomer, the monomers are blended with an inert organic solvent which is water miscible or water dispersible and a cobalt chain transfer agent and heated usually to the reflux

20 temperature of the reaction mixture. In subsequent steps additional monomers and cobalt catalyst and conventional polymerization catalyst are added and polymerization is continued until a macromonomer is formed of the desired molecular weight.

Preferred cobalt chain transfer agents or catalysts are

25 described in US Patent 4,680,352 to Janowicz et al and US Patent 4,722,984 to Janowicz. Most preferred are pentacyanocobaltate (II), diaquabis(borondifluorodimethyl-glyoximato) cobaltate(II) and diaquabis(borondifluorophenylglyoximato) cobaltate (II). Cobalt (III) versions of these catalysts are also preferred. Typically these chain transfer

30 agents are used at concentrations of about 5-1000 ppm based on the monomers used.

The macromonomer is preferably formed in a solvent or solvent blend using a free radical initiator and a Co (II) or (III) chelate chain transfer agent. Examples of solvents are aromatics, aliphatics, ketones, glycol ethers, acetates, alcohols as, e.g., methyl ethyl ketone, isopropyl alcohol, n-

butylglycolether, n-butyl diethyleneglycol ether, propylene glycol methylether acetate, propyleneglycol methylether, and N-butanol.

Peroxy- and azo-initiators (0.5-5% weight on monomer) can be used in the synthesis of the macromonomers in the presence of 2-5,000 ppm (on total monomer) or Co (II) chelate in the temperature range between 70-160°C, more preferably azo-type initiators as, e.g., 2,2'-azobis (2,4 dimethylpentanenitrile), 2,2'-azobis (2-methylpropanenitrile), 2,2'-azobis (2-methylbutanenitrile), 1,1'-azo (cyclohexane carbonitrile) and 4,4'-azobis (4-cyanopentanoic) acid.

After the macromonomer is formed as described above, solvent is optionally stripped off and the backbone monomers are added to the macromonomer along with additional solvent and polymerization catalyst. Any of the aforementioned azo-type catalysts can be used as can other suitable catalysts such as peroxides and hydroperoxides. Typical of such catalysts are di-tertiarybutyl peroxide, di-cumylperoxide, tertiaryamyl peroxide, cumenehydroperoxide, di(n-propyl) peroxydicarbonate, peresters such as amyl peroxyacetate and the like. Commercially available peroxy type initiators include, e.g., t-butylperoxide or Triganox™ B from AKZO, t-butylperacetate or Triganox™ FCS0 from AKZO, t-butylperbenzoate or Triganox™ C from AKZO, and t-butylperpivalate or Triganox™ 25 C-75 from AKZO.

Polymerization is continued at or below the reflux temperature of the reaction mixture until a graft copolymer is formed of the desired molecular weight.

Typical solvents that can be used to form the macromonomer or the graft copolymer are ketones such as methyl ethyl ketone, isobutyl ketone, ethyl amyl ketone, acetone, alcohols such as methanol, ethanol, isopropanol, esters such as ethyl acetate, glycols such as ethylene glycol, propylene glycol, ethers such as tetrahydrofuran, ethylene glycol mono butyl ether and the like.

In the synthesis of the macromonomer and/or the graft copolymer small amounts of difunctional alpha-beta unsaturated compounds can be used as, e.g., ethyleneglycol dimethacrylate or hexanedioildiacrylate.

After the graft copolymer is formed, it is neutralized with an amine or an inorganic base such as ammonium hydroxide or sodium hydroxide and then water is added to form a dispersion. Typical amines that

can be used include AMP (2-amino-2-methyl-1-propanol, dimethyl-AMP, amino methyl propanol, amino ethyl propanol, dimethyl ethanol amine, triethylamine and the like. One preferred amine is amino methyl propanol and the preferred inorganic base is ammonium hydroxide.

5 The conversion into a water dispersion may be accomplished preferably by stripping our 30 to 60% of the solvent followed by admixing with an organic amine or ammonia and diluting with water, or by admixing with a solution of water and amine after the solvent stripping. Alternatively, the polymerized graft copolymer solution, after stripping, can be stirred
10 slowly into a solution of water and the amine. The degree of neutralization of the dispersion can be from 10 to 150% of the total amount of acid groups, preferably from 40-105%. The final pH of the dispersion can accordingly be about 4-10, preferably 7-9. The solvents can be stripped-off eventually afterwards.

15 The overall graft copolymer water borne dispersion should be characterized by an acid value of from 5 to about 150 (mg KOH/g resin solids), more preferably from 10 to about 70 and still more preferably from 15 to about 35, and an hydroxyl number of about 0 to about 250 (mg KOH/g resin solids), more preferably from 40 to 150. If the graft copolymer
20 waterborne dispersion is used to be crosslinked with a water dispersable polyisocyanate, it could also have, but not necessarily, an amine number (mg KOH/g resin solids) of 0-200, more preferably from 20-150.

Particularly useful graft copolymers include the following:
25 a graft copolymer having a backbone of polymerized acrylate or methacrylate monomers, styrene monomers, methacrylic or acrylic acid monomers, and hydroxy-functional acrylate or methacrylate monomers, and side chains of a macromonomer having a weight average molecular weight of about 2,000-30,000 and containing about 50% by weight, based on the weight of the backbone, of polymerized alkyl methacrylate or acrylate monomers,
30 alkyl hexyl acrylate or methacrylate monomers, and hydroxy-functional acrylate or methacrylate monomers. based on the weight of the backbone, of polymerized methacrylic acid.

35 a graft copolymer having the above backbone of side chains comprising polymerized methyl methacrylate, butyl acrylate, methacrylic acid, styrene, and hydroxyethyl acrylate.

a graft copolymer having macromonomers comprising polymerized 2-ethylhexyl acrylate, butyl methacrylate, and hydroxyethyl methacrylate.

- 5 The afore described binder systems are utilized to produce waterborne coatings by blending with other suitable components in accordance with normal paint formulation techniques.

Linear film-forming polymers, preferably 0 to 55 percent by weight, based on the weight of the binder, may also be used in conjunction with the graft copolymer.

- 10 The graft copolymers of the present invention are useful as film forming vehicles in the preparation of high solids coating compositions such as, for example, clearcoat compositions useful in automotive applications.

- 15 In preparing the coating compositions of the present invention, the graft copolymer is combined with a crosslinking agent.

- The binder may be cured with a curing agent containing N-methylol and/or N-methylol ether groups or blocked isocyanate, as well be readily practiced by those skilled in the art. Examples of such curing agents are amino resins obtained by reacting an aldehyde, such as formaldehyde, 20 with a compound containing amino group such as melamine, urea and benzoguanamine and total or partial etherification of the N-methylol group with an alcohol such as, e.g., methanol, n-butanol, isobutanol.

- 25 It is possible to use the present branched copolymers in a lacquer without a cross-linker, particularly for ambient dried coating systems or automotive refinish.

- To form a composition which will crosslink under elevated baking temperatures of about 60-180°C for about 5-60 minutes, about 10 to 40%, preferably 15 to 30% by weight, based on the weight of the binder, of a water-soluble water dispersible alkylated melamine formaldehyde 30 crosslinking agent having 1-4 carbon atoms on the alkylated group is preferred.

- These crosslinking agents are generally partially alkylated melamine formaldehyde compounds and may be monomeric or polymeric and if polymeric have a degree of polymerization of about 1-3. Typical 35 alcohols used to alkylate these resins are methanol, ethanol, propanol, butanol, isobutanol, and the like. Preferred alkylated melamine crosslinking

agents that are commercially available include Cymel™ 373, Cymel™ 385, Resimine™ 714, Resinine™ 730 and 731, Resimine™ 735 and 745.

Coating compositions of this invention containing a melamine crosslinking agent can contain about 0.1 to 1.0%, based on the weight of a binder, of a strong acid catalyst or a salt thereof to lower curing temperatures and time. Paratoluene sulfonic acid is a preferred catalyst or its ammonium salt. Other catalysts that can be used are dodecyl benzene sulfonic acid, phosphoric acid and amine or ammonium salts of these acids.

As indicated above, other film forming polymers can also be used such as acrylourethanes, polyesters and polyester urethanes, polyethers and polyether urethanes that are compatible with the graft polymer. From about 20 to 70 percent, preferably 30 to 50 percent of a latex may also be used in conjunction with the above-described branched polymer. As well known by those skilled in the art, acrylic lattices are commercially available from ICI, Rohm & Haas and others, for example, Neocryl™ latex (ICI, Wilmington, DE).

Although the composition is aqueous, a solvent is also typically utilized, preferably in minimal amounts, to facilitate formulation and application of the coating compositions of the present invention. An organic solvent is utilized which is compatible with the components of the composition. The amounts of graft copolymer, curing agent, and catalyst will, of course, vary widely depending upon many factors, among them the specific components of the composition and the intended use of the composition.

In addition, a composition according to the present invention may contain a variety of other optional ingredients, including pigments, pearlescent flakes, fillers, plasticizers, antioxidants, surfactants and flow control agents.

To improve weatherability of a finish produced by the present coating composition, an ultraviolet light stabilizer or a combination of ultraviolet light stabilizers can be added in the amount of about 0.1-5% by weight, based on the weight of the binder. Such stabilizers include ultraviolet light absorbers, screeners, quenchers, and specific hindered amine light stabilizers. Also, an antioxidant can be added, in the about 0.1-5% by weight, based on the weight of the binder.

Typical ultraviolet light stabilizers that are useful include benzophenones, triazoles, triazines, benzoates, hindered amines and mixtures

thereof. Specific examples of ultraviolet stabilizers are disclosed in U.S. Patent 4,591,533, the entire disclosure of which is incorporated herein by reference.

The composition may also include conventional formulation additives such as flow control agents, for example, Resiflow® S (polybutylacrylate), BYK 320 and 325 (high molecular weight polyacrylates); rheology control agents, such as fumed silica and thickeners such as the Acrylsol™ copolymers from Rohm & Haas.

When the present composition is used as a clearcoat (topcoat) over a pigmented colorcoat (basecoat) to provide a colorcoat/clearcoat finish, small amounts of pigment can be added to the clear coat to provide special color or aesthetic effects such as tinting.

The composition has excellent adhesion to a variety of metallic or non-metallic substrates, such as previously painted substrates, cold rolled steel, phosphatized steel, and steel coated with conventional primers by electrodeposition. The present composition can be used to coat plastic substrates such as polyester reinforced fiberglass, reaction injection-molded urethanes and partially crystalline polyamides.

When the present coating composition is used as a basecoat, colorcoat, monocoat, primer surfacer or primer, then the composition includes pigments such as, for example, the following: metallic oxides such as titanium dioxide, zinc oxide, iron oxides of various colors, carbon black, filler pigments such as talc, china clay, barytes, carbonates, silicates and a wide variety of organic colored pigments such as quinacridones, copper phthalocyanines, perylenes, azo pigments, indanthrone blues, carbazoles such as carbozole violet, isoindolinones, isoindolones, thioindigo reds, benzimidazolinones, metallic flake pigments such as aluminum flake and the like.

The pigments can be introduced into the coating composition by first forming a mill base or pigment dispersion with any of the aforementioned polymers used in the coating composition or with another compatible polymer or dispersant by conventional techniques, such as high speed mixing, sand grinding, ball milling, attritor grinding or two roll milling. The mill base is then blended with the other constituents used in the coating composition to obtain the present coating compositions.

The coating composition can be applied by conventional techniques such as spraying, electrostatic spraying, dipping, brushing, flowcoating and the like. The preferred techniques are spraying and electrostatic spraying. In OEM applications, the composition is typically 5 baked at 100-150°C for about 15-30 minutes to form a coating about 0.1-3.0 mils thick. When the composition is used as a clearcoat, it is applied over the colorcoat which may be dried to a tack-free state and cured or preferably flash dried for a short period before the clearcoat is applied. The colorcoat/clearcoat finish is then baked as mentioned above to provide a 10 dried and cured finish. The present invention is also applicable to non-baking refinish systems, as will be readily appreciated by those skilled in the art.

It is customary to apply a clear topcoat over a basecoat by means of a "wet-on-wet" application, i.e., the topcoat is applied to the 15 basecoat without curing or completely drying the basecoat. The coated substrate is then heated for a predetermined time period to allow simultaneous curing of the base and clear coats.

The following Examples illustrate the invention. All parts and percentages are on a weight basis unless otherwise indicated. All molecular 20 weights disclosed herein are determined by gel permeation chromatography using a polystyrene standard.

EXAMPLE 1

This example illustrates the preparation of a macromonomer 25 of 5% isobutyl methacrylate, 10% hydroxyethyl methacrylate, and 85% 2-ethyl hexyl methacrylate, for use in preparing a branched copolymer. To a 2-liter flask fitted with an agitator, condenser, heating mantle, nitrogen inlet, thermocouple and an addition port was added 20.32 g of isobutyl methacrylate, 345.5 g of 2-ethyl hexyl methacrylate, 40.64 g of hydroxy ethyl 30 methacrylate and 183.8 g of toluene. The mixture was agitated and heated to reflux (122-126° C) under nitrogen. To this was then added, as a shot, a pre-mix of a solution of 0.35 g of Vazo® 88 initiator, 13.8 g of toluene and 12.9 g of a 0.17% solution of bis(boron difluoro diphenyl glyoximato) cobaltate(II) 35 in methyl ethyl ketone. This was followed by the addition of a pre-mix of a solution of 17.8 g of isobutyl methacrylate, 303.2 g of 2-ethyl hexyl methacrylate, 35.7 g of hydroxy ethyl methacrylate, 1.35 g of Vazo® 88

initiator, and 12.0 g of xylene, 74.5 g of toluene and 4.3 g of a 0.17% solution of bis(boron difluoro diphenyl glyoximato) cobaltate(II) in methyl ethyl ketone over 240 mins. while maintaining reflux (116-122° C). Following a 30 min. hold period, a pre-mixed solution of 0.32 g of Vazo® 88 initiator and 5 30.54 g of toluene was added over 60 mins. while maintaining reflux. The batch was then held at reflux for an additional 60 mins. and then cooled. The macromonomer thus prepared has a number average molecular weight of 6469 and a weight average molecular weight of 9546 as determined by GPC. Weight solids are 63.6% and Gardner viscosity I. The percent terminal vinyl 10 unsaturation is greater than 95 as determined by thermogravimetric analysis.

EXAMPLE 2

This example illustrates the preparation of a branched copolymer for use in a composition according to the present invention. To a 5-liter flask fitted with an agitator, condenser, heating mantle, nitrogen inlet, thermocouple and an addition part was added 1490.1 g of macromonomer from Example 1 above, 457.8 g of Buty Cellosolve® carrier, 305.2 g of isopropanol and the temperature raised to reflux (100-105° C) under nitrogen. This was followed by the addition of a premixed solution of 381.5 g 15 of methyl methacrylate (MMA), 47.7 g of methacrylate acid (MAA), 190.7 g of styrene (STY), 95.4 g of hydroxy ethyl acrylate (HEA), 238.4 g of butyl acrylate (BA), 11.92 g of Vazo® 88 initiator and 95.4 g of Butyl Cellosolve® carrier and 47.7 g of isopropanol over 180 minutes holding temperature at reflux. Following a 60 minute hold period, a premixed solution of 4.77 g of 20 Vazo® 88 initiator and 133.51 g of toluene was added over 60 min holding reflux. This was followed by a hold period of 120 min at reflux after which the batch was cooled. Molecular weight via gel permeation chromatography is 8946 number average and 20830 weight average. Weight solids are 55% 25 and Gardner viscosity is R. The ration of backbone to macromonomer arms is about 50/50. The composition of the backbone is 30 STY/MMA/BA/HEA/MAA in the weight ratio of 20/40/25/10/5.

EXAMPLE 3

This example illustrates the preparation of an aqueous dispersion of a branched copolymer. To a 2-liter flask fitted with an agitator, condenser, heating mantle, nitrogen inlet, thermocouple and an addition port 35

was added 500 g of branched copolymer from Example 2 above and the temperature raised to distill 133.3 g of solvent. The batch was cooled to 110° C at which time a mixture of 4.27 g dimethyl ethanol amine and 10 g deionized water were added to neutralize the acid functionality of the branch 5 copolymer to 60%, theoretical. The batch was agitated for 10 mins. at which time 719.1 gms. deionized water was slowly added with good agitation and the temperature reduced to less than 40° C. A white dispersion of the branched polymer was obtained of weight solids 25%, Gardner viscosity A3, PH 8.7 and a particle size of 147 nanometers as determined by quasielectric 10 light scattering.

EXAMPLE 4

This example illustrates another preparation of an aqueous dispersion of a branched copolymer according to the present invention. A 15 process identical to Example 3 was carried out except that the level of amine is 5.69 g to achieve a percent neutralization of 80 percent. Weight solids are 25%, Gardner viscosity A2, Ph 9.05 and a particle size of 100 nanometers.

EXAMPLE 5

20 This example illustrates a composition prepared according to Example 3 except with a level of amine of 7.11 g to achieve a percent neutralization of 100 percent. Weight solids are 25%, Gardner viscosity A + 1/2, pH 9.20 and a particle size of 71 nanometers.

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EXAMPLE 6

This example illustrates the preparation of a macromonomer for use in the present invention. To a 2-liter flask fitted with an agitator, condenser, heating mantle, nitrogen inlet, thermocouple and an addition port was added 20.3 g of isobutyl methacrylate, 304.8 g of 2-ethyl hexyl 30 methacrylate, 81.4 g of hydroxy ethyl methacrylate, 160.3 g of toluene and 100.3 g of butyl acetate. The mixture was agitated and heated to reflux (135-140°C) under nitrogen. To this was then added, as a shot, a pre-mix of a solution of 0.35 g of Vazo® 88 initiator, 13.8 g of toluene and 17.2 g of a 0.17% solution of bis(boron difluoro diphenyl glyoximato) cobaltate(II) in 35 methyl ethyl ketone. This was followed by the addition of a pre-mix of a solution of 17.8 g of isobutyl methacrylate, 268.1 g of 2-ethyl hexyl

methacrylate, 71.6 g of hydroxy ethyl methacrylate, 1.35 g of Vazo 88 initiator and 86.8 g of toluene over 240 mins. while maintaining reflux (116-122°C). Following a 30 min hold period, a pre-mixed solution of 0.32 g of Vazo® 88 initiator and 23.0 g of toluene was added over 60 min while

5 maintaining reflux. The batch was then held at reflux for an additional 60 min at which time a mixture of 0.23 g of t-butyl peroctoate and 31.5 g of Butyl Cellosolve® carrier was added as a shot and the reaction mixture then cooled. The macromonomer thus prepared has a number average molecular weight of 5250 and a weight average molecular weight of 8920 as determined

10 by GPC. Weight solids are 63.6% and Gardner viscosity T. The percent terminal vinyl unsaturation is greater than 95 as determined by thermogravimetric analysis. The composition of the macromonomer is 75 parts 2-ethyl hexyl methacrylate, 5 parts isobutyl methacrylate and 20 parts hydroxy ethyl methacrylate.

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EXAMPLE 7

This example illustrates the preparation of a branched copolymer for use in the present invention. To a 2-liter flask fitted with an agitator, condenser, heating mantle, nitrogen inlet, thermocouple and an addition port was added 306.5 g of macromonomer from Example 6 above, 272.8 g of Butyl Cellosolve® carrier, 22.7 g of toluene and the temperature raised to 100°C under nitrogen. This was followed by the addition of a premixed solution of 122.8 g of methyl methacrylate, 31.8 g of methacrylic acid, 91.0 g of styrene, 91.0 g of hydroxy ethyl acrylate, 118.2 g of butyl acrylate, 11.4 g of Vazo® 88 carrier, 9.1 g of Butyl Cellosolve® carrier and 77.3 g of toluene over 180 min holding the temperature at 100°C. This was followed by a hold period of 60 min at 100°C at which time the temperature was lowered to 90 °C over 30 min and the batch held at this temperature with agitation for 270 min and then cooled. Molecular weight via gel permeation chromatography is 16040 number average and 39470 weight average. The weight solids are 55% and Gardner viscosity is Y. The composition of the backbone is STY/MMA/BA/HEA/MAA in the weight ratio of 20/27/26/20/7 and the ratio of backbone to the graft is 70/30 by weight.

EXAMPLE 8

This example illustrates the preparation of a water-based dispersion 80 percent neutralized. To a 2-liter flask fitted with an agitator, condenser, heating mantle, nitrogen inlet, thermocouple and an addition port
5 was added 500 g of branched copolymer from Example 7 above and the temperature raised to distill 133 g of solvent. The batch was cooled to 80 °C at which time 11.6 g AMP-95® (95% solution of 2-amino 1 propanol in water) was added to neutralize the acid functionality of the branch copolymer to 80%, theoretical. The batch was agitated for 10 mins. at which
10 time 722 g of deionized water was slowly added with good agitation and the temperature reduced to less than 40°C. A white dispersion of the branched copolymer was obtained of weight solids 25%, Gardner viscosity L, pH 8.5 and a particle size of 29 nanometers as determined by quasielectric light scattering.

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EXAMPLE 9

This example illustrates the preparation of a water-based dispersion neutralized with another amine compound. A process identical to Example 8 was run substituting dimethyl amino ethanol on an equivalent basis for AMP. The pH of the dispersion in water was 8.7, viscosity M and particle size was 34 nanometers.

EXAMPLE 10

This example illustrates the preparation of a water-based dispersion neutralized with another amine compound A process identical to Example 8 was run substituting diisopropanol amine on an equivalent basis for AMP. The pH of the dispersion in water was 8.4, viscosity J and particle size was 35 nanometers.

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EXAMPLE 11

This example illustrates the preparation of a water-based dispersion neutralized with ammonia. A process identical to Example 8 was run substituting ammonia on an equivalent basis for AMP. The pH of the dispersion in water was 8.1, viscosity A and particle size was 45 nanometers.

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EXAMPLE 12

This example illustrates the preparation of a branched copolymer. To a 5-liter flask fitted with an agitator, condenser, heating mantle, nitrogen inlet, thermocouple and an addition port was added 1941.2 g of macromonomer from Example 6 above, 472.1 g of Butyl Cellosolve® carrier and the temperature raised to 110°C under nitrogen. This was followed by the addition of a premixed solution of 165.7 g of methyl methacrylate, 58.0 g of acrylic acid, 331.0 g of styrene, 82.8 g of hydroxy ethyl acrylate, 190.5 g of butyl acrylate, 26.5 g of Vazo® 88 initiator, 74.5 g of Butyl Cellosolve® carrier, 107.7 g of methyl ethyl ketone and 49.7 g of toluene over 180 min holding the temperature at 110 °C. This was followed by a hold period of 180 min at 110°C at which time the temperature was cooled. The molecular weight via gel permeation chromatography was 8720 number average and 20570 weight average. The weight solids are 60% and the Gardner viscosity is U. The composition of the backbone is STY/MMA/BA/HEA/AA in the weight ratio of 40/20/23/10/7 and the weight ratio of backbone to graft is 40/60.

EXAMPLE 13

20 This example illustrates the preparation of another water-based dispersion. To a 5-liter flask fitted with an agitator, condenser, heating mantle, nitrogen inlet, thermocouple and an addition prot was added 1397 g of branched copolymer from Example 12 and the temperature raised to distill 351 g of solvent. The batch was cooled to 80°C at which time 20 g of 25 dimethyl ethanol amine was added to neutralize the acid functionality of the branch copolymer to 70%, theoretical. The batch was agitated for 10 mins. at which time 1724 g of deionized water was slowly added with good agitation and the temperature reduced to less than 40°C. A white dispersion of the branched polymer was obtained of weight solids 30%, Gardner viscosity T, 30 pH 8.3 and a particle size of 321 nanometers as determined by quasielectric light scattering.

EXAMPLE 14

35 This example illustrates the preparation of an acrylic latex polymer for use in the present composition. To a 5-liter flask fitted with an agitator, condensor, heating mantle, nitrogen inlet, thermocouple and an

addition port was added 1308 g deionized water, and the temperature raised to 85°C with agitation. In a separate container, the following mixture (Part 1) was prepared: 229 g methyl methacrylate, 160 g styrene, 875 g of 2-ethyl hexyl methacrylate, 320 g hydroxyethyl methacrylate, 16 g methacrylic acid, 5 20 g Trem™ LF-40 (Henkel, Inc.) and 961 g deionized water. This mixture was preemulsified with an Eppenbach™ homogenizer and 5% was then added to the flask after an 85°C temperature was attained. This was followed by addition, as a shot, of a mixture of 30 g deionized water and 3 g ammonium persulfate. The remainder (95%) of Part 1 was then added to 10 the flask over 90 min keeping the temperature constant at 88-90°C. The addition of Part 1 was followed by the addition, as a shot, of a mixture of 10 g deionized water and 1 g ammonium persulfate after which the contents of the flask were held at 88-90°C for 2 hours with agitation. The mixture was cooled to 60°C at which time 50g of water and 17 g of 2-amino-1-propanol 15 was added slowly and the mixture further cooled to room temperature. The percent weight solids are 40 and the particle size of the latex is 200 nanometers (nm).

EXAMPLE 15

20 This example illustrates the preparation of a clearcoat formulation according to the present invention. To a pint can was added, with stirring, 200 g of aqueous branched polymer dispersion from Example 2 above, 15.4 g Cymel™325 (polymeric melamine from American Cyanamid), 2.2 g of a 28% solution of dodecyl benzene sulfonic acid and 9% AMP-95™ 25 in deionized water, 5.6 g ASE-60™ thickener (from Rohm & Haas) and 5.6 g of a 10% solution of dimethyl ethanol amine in deionized water. The mixture was agitated for 30 min. The pH is 8.5 and the 5-rpm-Brookfield viscosity is 12 poise. A drawdown of the clear was applied over Elpo™ primed-steel panels with an 8 milblade and cured at 265°F for 30 min. Dry- 30 film thickness is 1.8 mils. This clearcoat was smooth, hard (11 Knoop), well cured (100+ MEK solvent rubs) and passed a 96 hour humidity test (110°F/100% humidity) with no blistering or water spotting.

EXAMPLE 16

35 This example illustrates a basecoat formulation according to the present invention. The procedure and ingredients were the same as in

Example 15, but with the addition of 82 g of Ti Pure R-942™ titanium oxide slurry (76%) in deionized water (from DuPont). The paint was adjusted to 12 poise viscosity (5-rpm Brookfield) with 16 g water and spray applied onto Elpo™ primed-steel panels, followed by a 5-min bake at 165°F. This
5 basecoat was then clearcoated with an acrylic/isocyanate (DES™ 3390) high-solids clearcoat, followed by a bake of 30 min at 250°F. The dry-film thickness of the basecoat was 1.4 mils and that of the clearcoat was 1.8 mils. The basecoat/clearcoat was glossy (20° gloss of 94), smooth (95 distinctness of image or DOI), hard, and passed a 96-hour humidity test with no blistering
10 or water spotting.

EXAMPLE 17

This example illustrates another clearcoat formulation according to the present invention. To a pint can was added with stirring 90
15 g of aqueous branched polymer from Example 8 above, 74 g of latex from Example 14 above, 12.1 g Cymel™ 1161 polymeric melamine (from American Cyanamid), 2.2 g of a 28% solution of dodecyl benzene sulfonic acid and 9% AMP-95™ in deionized water, 21.5 g butyl Cellosolve™ and 5 g of a 10% solution of AMP-95™ in deionized water. This mixture was
20 agitated for 30 min and reduced to a viscosity of 12 poise at 5-rpm Brookfield. A drawdown of the clear was applied over Elpo™ primed steel panels with an 8 miliblade and cured at 265°F for 30 min. The dry-film thickness was 1.8 mils. This clearcoat was smooth, hard, and passed the 96-hour humidity test (110°F/100% humidity) with no blistering or water
25 spotting.

EXAMPLE 18

This example illustrates another basecoat formulation according to the present invention. To a pint can was added, with stirring,
30 456 g of aqueous branched polymer dispersion from Example 2 above, 52 g Cymel™ 325 polymeric melamine (form American Cyanamid), 7.3 g of a 28% solution of dodecyl benzene sulfonic acid and 9% AMP-95™ solution in deionized water, 4 g butyl Cellosolve™, 173 g Spensol™ L-52 polyurethane
35 (from Reichold Chemical) and 273 g of Ti Pure R-942™ TiO₂ slurry (76%) in deionized water (from DuPont). The pH was 8.5 and the viscosity (5-rpm Brookfield) was 12 poise. The basecoat was sprayed over Elpo™.

primed-steel panels, baked at 165°F for 5 min and then clearcoated with an acrylic/isocyanate (DES™ 3390) clearcoat. The basecoat/clearcoat combination was then cured at 250°F for 30 min. The final coating was hard, glossy and passed a 96-hour humidity test with no blistering or water spotting.

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Various modifications, alterations, additions or substitutions of the components of the compositions of this invention will be apparent to those skilled in the art without departing from the scope and spirit of this invention. This invention is not limited to the illustrative embodiments set forth herein, but rather the invention is defined by the following claims.

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CLAIMS

We claim:

1. A waterbased coating composition comprising an aqueous medium having dispersed therein a graft copolymer comprising 5-95% by weight of an acid-functional acrylic backbone and correspondingly 95-5% by weight of macromonomer side chains; wherein
 - the side chains consist essentially of macromonomers of polymerized ethylenically unsaturated monomers that are
 - 10 polymerized into the backbone via an ethylenically unsaturated group and the macromonomers have a weight average molecular weight of about 1,000-30,000 and
 - 15 the backbone consists essentially of polymerized ethylenically unsaturated monomers and 2-30% by weight, based on the weight of the backbone of polymerized ethylenically unsaturated monomers having an acid-functional group; and wherein at least 10% of the acid-functional groups are neutralized with an amine or an inorganic base.
2. The composition of claim 1, wherein the graft polymer comprises backbone 30 to-70 percent by weight of the backbone and 70 to 30 percent by weight of the side chains.
3. The composition of claim 1 wherein the backbone comprises 3-15% by weight of monomers which have a carboxylic acid functionality or salt thereof.
4. The composition of claim 1, wherein the graft copolymer comprises 5-40% by weight of polymerized monomers which have a hydroxy functionality.
- 30 5. The composition of claim 1, further comprising a melamine crosslinker.
6. A coating composition, useful for coating a substrate, which 35 composition comprises:

- (a) 10 to 90%, by weight of the composition, based on the weight of the binder, of a graft copolymer having a molecular weight of 2000 to 100,000 comprising: a graft copolymer comprising 5-95% by weight of an acid-functional acrylic backbone and correspondingly 95-5% by weight of a macromonomer side chains; wherein
- 5 (i) the side chains consist essentially of macromonomers of polymerized ethylenically unsaturated monomers that are polymerized into the backbone via an ethylenically unsaturated group and the macromonomers have a weight average molecular weight of about 1,000-30,000; and
- 10 (ii) the backbone consists essentially of polymerized ethylenically unsaturated monomers and 2-30% by weight, based on the weight of the backbone of polymerized ethylenically unsaturated acid monomers; at least 10% of the acid groups
- 15 are neutralized with an amine or an inorganic base; and
- (b) 10 to 40%, by weight of the binder, of a crosslinking agent; and
- (c) from 10 to 90% by weight, based on the weight of the composition, of an aqueous carrier.
- 20 7. The composition of claim 1 in which the backbone contains 2-30% by weight of an acid functional monomer and wherein the graft polymer has a weight average molecular weight of 5000 to 40,000, an acid value of 30-700, an amine value of 0-200, and a hydroxyl value of 0-300.
- 25 8. The composition of claim 1 wherein the final graft copolymer has preferably a hydroxyl value of 30-150, an amine value of 0-100, an acid value of 10-60 and a weight average molecular weight of 5000-100,000.
- 30 9. The composition of claim 6, wherein said backbone or macromonomers further comprises polymerized ethylenically unsaturated monomers not containing carboxylic functionality, which monomers are selected from the group consisting of alkyl acrylates, alkyl methacrylates, cycloaliphatic acrylates, cycloaliphatic methacrylates, aryl acrylates, aryl methacrylates, styrene, alkyl styrene, acrylonitrile, and mixtures thereof; and
- 35

wherein the ethylenically unsaturated monomers containing carboxylic functionality comprise monomers selected from the group consisting of carboxylic alkyl acrylates and carboxylic alkyl methacrylates, wherein the above-mentioned alkyl, cycloaliphatic, and aryl groups have 1 to 12 carbon atoms.

10. A process for preparing a coating composition in which the principal polymer is a graft copolymer comprising a polymeric backbone having a plurality of macromonomers attached thereto, the backbone having carboxylic functionalities capable of reacting with a crosslinking polymer, the process comprising
- (a) preparing macromonomers, in an organic solvent, by polymerizing, using a catalytic chain transfer agent containing cobalt, ethylenically unsaturated monomers comprising ethylenically unsaturated monomers ; and
- (b) forming, in an organic solvent, the graft copolymer by polymerizing, in the presence of said macromonomers prepared in step (a), ethylenically unsaturated monomers containing 2 to 30% by weight of carboxylic acid groups, whereby said macromonomers are incorporated into said backbone at a single terminal point of said macromonomers, such attachment to said backbone occurring by the reaction of a terminal ethylene unsaturation on each of said macromonomers with monomers which polymerize to form said backbone, thereby forming a graft copolymer with a molecular weight of 2000-100,000, which after neutralizing with an amine can be dispersed in water.

1
INTERNATIONAL SEARCH REPORT

International application No.

PCT/US 95/00376

A. CLASSIFICATION OF SUBJECT MATTER

IPC6: C08F 290/04, C09D 151/00

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC6: C08F, C09D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

WPIL

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	US, A, 5010140 (JOSEPH A. ANTONELLI ET AL), 23 April 1991 (23.04.91) --	1-10
A	WO, A1, 9303081 (E.I. DU PONT DE NEMOURS AND COMPANY), 18 February 1993 (18.02.93) -----	1-10

 Further documents are listed in the continuation of Box C. See patent family annex.

- * Special categories of cited documents:
- "A" document defining the general state of the art which is not considered to be of particular relevance
- "E" earlier document but published on or after the international filing date
- "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- "O" document referring to an oral disclosure, use, exhibition or other means
- "P" document published prior to the international filing date but later than the priority date claimed
- "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
- "Z" document member of the same patent family

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Name and mailing address of the International Searching Authority

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INTERNATIONAL SEARCH REPORT

SA 103603

Information on patent family members

01/04/95

International application No.
PCT/US 95/00376

Patent document cited in search report	Publication date	Patent family member(s)		Publication date
US-A- 5010140	23/04/91	AU-B-	627419	20/08/92
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